

ISCO reagent (sodium persulfate and calcium peroxide)



Chemical reagent used in ISCO processes and aerobic bioremediation in soils and groundwater that uses sodium persulfate and calcium peroxide as active components. The oxidant mixture enables short-term in situ chemical oxidation (ISCO) and provides electron acceptors for long-term biological oxidation.

Features

Recent applications of in situ chemical oxidation (ISCO) have shown this remedial strategy to be cost effective for recalcitrant organic contaminants in groundwater and soils. The application of sodium persulfate and calcium peroxide has a double action mechanism, being used as an ISCO reagent and oxygen supply in aerobic

bioremediation processes. The mixture of sodium persulfate and calcium peroxide provides one of the strongest chemical oxidants for ISCO in the short term, and also provides electron acceptors (oxygen and sulphate) for long-term biological oxidation.

PARAMETER	TYPICAL VALUE	UNITS
Formula	Na2S2O8 + CaO2	-
Appearance	Odourless white granular solid	-
Specific gravity	2,60 – 2,92	g/mL
Melting point	275	оС
рН	11,3 – 12,1	-
Water solubility	73 % soluble + 18% non-soluble	-



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Applications

The combined reagent sodium persulfate and calcium peroxide is used for the on-site treatment of groundwater petroleum contaminants by chemical oxidation and aerobic bioremediation. These types of contaminants include BTEX

(benzene, toluene, ethylbenzene, and xylene), MTBE (methyl tert-butyl ether), PAH (polycyclic aromatic hydrocarbons), chlorinated alkenes, chlorinated alkanes, 1,4-dioxane, PCBs (polychlorinated biphenyls), and pentachlorophenol.

Advantages

- Supports chemical oxidation and bioremediation processes.
- Works on a wide range of contaminants.
- Oxygen content around 46% by weight.
- Typically, 40% by weight solubility for persulfate.

Warnings and recommendations on prevention and safety

The reagent must be handled with care. Protective equipment during handling should include face shields and / or goggles, rubber or plastic gloves, and rubber or plastic apron. If clothes look stained, wash immediately; spontaneous ignition can occur with cloth or paper. In cases of significant exposure, use the appropriate NIOSH-MSHA dust or mist respirator. For more details do not hesitate to consult our SDS.

Storage

Avoid contact with acids, peroxides, and all combustible or easily oxidizable organic materials, including inorganic oxidizable materials and metal powders. With hydrochloric acid, chlorine gas is released. Liquid sodium permanganate is not combustible but will support combustion. It can decompose if exposed to intense heat. Fires can be controlled and extinguished by using large amounts of water. For more details do not hesitate to consult our SDS.

Delivery format	
19L drum.	

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